

S.No	Date	Class	Name of student	Topic	Sign
1.	11/11/19	BSC I (M)	Sangeeta		
2.	11/11/19	BSC I (M)	Dimple		
3.	13/11/19	BSC I (M)	Deepika	Chemical kinetics	Deepika
4.	13/11/19	BSC I (M)	Kajal Devi	order of Reaction	Kajal Devi
5.	13/11/19	BSC I (M)	Bhumika	second order of reaction	Bhumika
6.	20/11/19	BSC III (M)	Hemendra	Third law of thermodynamics	हेमंड
7.	20/11/19	BSC III (M)	Pushpanjali	co-ordination compound	पुष्पा
8.	20/11/19	BSC IV (M)	Hulasi	principles of Hulsni wauner.	Hulasi
9.	22/11/19	BSC I (B)	Vaishali Sahu	Zero order Reaction	VS
10.	22/11/19	BSC I (B)	Sharda Pandey	Arrhenius equation half life period	Sharda

11)	05/12/19	BSc-III(BI)	virendrakumar	HSAB	Vipatel
12)	05/12/19	BSc II(B)	chaudani	Carboxylic acid	chaudani
13)	09/12/19	BSc II(B)	Umeshwar	Bioinorganic (transition element)	Umeshwar
14)	09/12/19	BSc III (CM)	Chander	Gibbs Helmholtz (thermody)	Chander
15)	09/12/19	BSc III (CM)	Nimit	magnetic properties	Nimit

Sign of Principal

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